

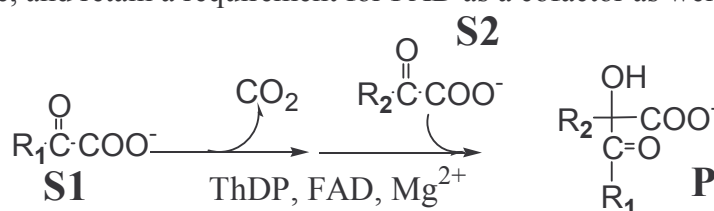
## Acetohydroxyacid Synthase and related decarboxylase-carboligases: origins of their substrate specificities

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Acetohydroxyacid synthases (AHAS) and glyoxylate carboligase (GCL) are thiamindiphosphate (ThDP) -dependent enzymes which normally catalyze decarboxylation of 2-oxoacids and condensation of the resulting hydroxyalkylThDP anions/enamines with a second oxoacid to form 2-acyl-2-hydroxyacids. These enzymes probably evolved from a pyruvate oxidase, and retain a requirement for FAD as a cofactor as well as for ThDP.



At least three different types of structural and mechanistic factors determine the varied specificities of these enzymes. In AHASs, pyruvate ( $\text{R}_1$ =methyl) is preferred by >10 fold over any other oxoacid as first substrate ( $\text{S}_1$ ) (2). Simple steric hindrance is the major determinant of this specificity, as a mutant AHAS with a strategically-placed Val→Ala mutation (*e.g.*, *E. coli* isozyme II V375A) allows 2-oxobutyrate ( $\text{R}_1$ =ethyl) to be a good first substrate and can synthesize propionhydroxybutyrate (**P**,  $\text{R}_1 = \text{R}_2 = \text{C}_2\text{H}_5$ ). An Ile residue in the position in GCL equivalent to the Val375 in AHAS II plays the analogous role in restricting wild-type GCL to glyoxylate as  $\text{S}_1$ .

GCL ( $\text{R}_1 = \text{H}$ ) is highly unusual among ThDP-dependent enzymes in that it has no carboxylate residue in H-bonding distance of the aminopyrimidine N1' of ThDP (3). Deprotonation at N2 of the thiazolium is apparently possible because of a very low dielectric constant in the vicinity of the thiazolium, lowering the  $\text{pK}_a$  for the thiazolium cation  $\Rightarrow$  ylide ionization. Glyoxylate is also more reactive to nucleophilic substitution than a ketoacid. The selective advantage of *loss* of the canonical activating Glu in GCL seems to be related, *inter alia*, to its effect on product release

The specificity of these enzymes for 2-oxoacids as  $\text{S}_2$  is due to an arginine residue which probably interacts with the carboxylate of  $\text{S}_2$  (*e.g.*, Arg276 in AHAS II) (1). Mutants altered at this arginine can utilize aromatic aldehydes as  $\text{S}_2$  and form chiral arylacyl carbinols, of interest as chiral synthons in pharmaceutical syntheses.

NMR measurements of the distribution of ThDP-bound intermediates reveals that faster rates of *release* of **P** when  $\text{R}_2 = \text{ethyl}$  compared to  $\text{R}_2 = \text{methyl}$  plays the major role in product specificity in AHASs (4). The crucial role of a Trp residue (Trp 464 in AHAS II) in determining this specificity may be due to control of a conformational change involved in product release rather than to affinity for 2-ketobutyrate. It is significant that in AHAS I, without the required Trp and with a low specificity for  $\text{R}_2 = \text{ethyl}$ , the product release step is readily reversible.

(1) Engel, S.; Vyazmensky, M.; Vinogradov, M. *et al. J Biol Chem* **2004**, 279, 24803-24812.

(2) Gollop, N.; Barak, Z.; Chipman, D. M. *Anal. Biochem.* **1987**, 160, 323-331.

(3) Kaplun, A.; Binshtein, E.; Vyazmensky, M. *et al. Nat Chem Biol* **2008**, 4, 113-118.

(4) Tittmann, K.; Vyazmensky, M.; Hubner, G. *et al. Proc Natl Acad Sci U S A* **2005**, 102, 553-558.